

Perchlorate Quantitation in Drinking Water Using Suppressed Anion Chromatography Coupled with Single Quadrupole MS

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1. Introduction

Despite current lack of regulation at Federal level, there are several states where perchlorate needs to be monitored in drinking water to ensure compliance with the corresponding Maximum Contaminant Levels. A few perchlorate determination methods have been developed by the U.S. EPA. EPA Method 332.0 is one of the most sensitive methods using suppressed ion chromatography coupled with mass spectrometry.

In this study, an IC-MS method was developed using a novel electrolytically regenerated anion suppressor as a part of a modular IC system coupled with a LC-MS 2020 single quadrupole MS to achieve higher sensitivity and selectivity for perchlorate determination in drinking water. Both conductivity detection (CDD) and MS with electrospray ionization are used in the method. The CDD is used to monitor the column effluent, only the perchlorate fraction is directed to MS via a divert valve. Two predominant perchlorate ions are $^{35}\text{Cl}^{16}\text{O}_4^-$ and $^{37}\text{Cl}^{16}\text{O}_4^-$ with m/z are 99 and 101, respectively. Selected ion monitoring (SIM) of both ions was used in the method. The system produced linear response with coefficient of determination greater than 0.9999 from concentration of 0.25 ppb to 20 ppb. This study demonstrated that the robust and reliable performance of the novel modular IC-MS for the determination of perchlorate in drinking water.

2. Experimental

Experiments were performed using a modular Shimadzu IC system with built-in electrolytically regenerated suppressor and a single quadrupole mass spectrometer detector (LCMS-2020) with electrospray ionization (ESI) interface.

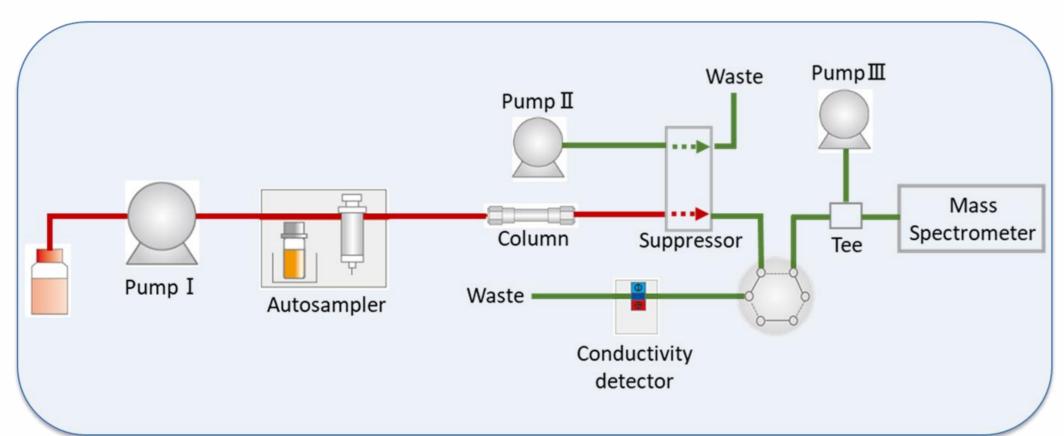
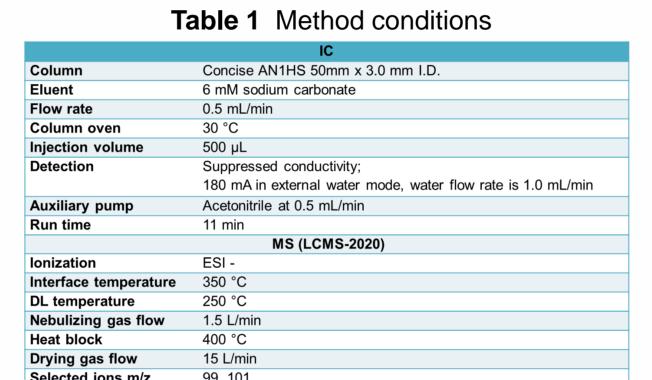


Fig. 1 Schematic diagram of IC-MS system for perchlorate analysis.

The schematic diagram of the IC-MS system used in this study is illustrated in Figure 1. In this method, external fresh water is continuously pumped through the suppressor regeneration channel at a 1 mL/min flow rate using the second pump to affect the electrolysis regeneration of the suppressor and yield a more stable, low noise baseline for the determination of perchlorate at low ppb levels. The conductivity detector is used to monitor column effluent. Only the perchlorate peak fraction is directed to the MS using a divert valve; this protects the MS from a buildup of trace salts from the eluent post-suppressor. A third pump provided an auxiliary flow of acetonitrile post-suppressor to enhance MS perchlorate response. Method conditions are listed in Table 1.



3. Results and Discussion

A five-point calibration curve was established across the concentration range of 0.25 to 20 ppb. Correlation coefficients of $r^2 > 0.9999$ were obtained for both ions, m/z 99 and 101, as shown in Figure 2.

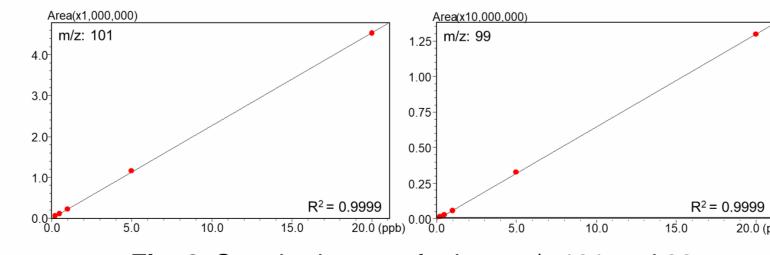


Fig. 2 Standard curves for ions m/z 101 and 99.

Method precision was performed using two perchlorate standards with concentration of 0.5 ppb and 1 ppb, respectively. Table 2 shows retention time precision and peak area precision of m/z 101 from 7 injections for each concentration. As shown in the table, excellent reproducibility was achieved for both retention time and peak area. Table 2 also shows the area count ratio of m/z 99 to 101 are 2.74 and 2.65 for 0.5 and 1 ppb perchlorate standard, respectively. The results shown in the table meet the area count ratio

requirement for confirmation of perchlorate, which is the range of 2.31-3.85 specified in EPA 332.0.

 Table 2 Retention time and peak area reproducibility over 7 injections

CIO ₄ - Standard (ppb)	Tr Precision (%RSD)	Area Precision (%RSD)	Area count ratio m/z 99 to 101
0.5	0.25	3.79	2.74
1.0	0.06	0.92	2.65

The performance and accuracy of the method were investigated by spiking perchlorate standards of 2 ppb and 10 ppb concentration into blank water. 96% recovery was obtained for the first concentration. A recovery of 99.3% was obtained for the second concentration.

Figure 3 shows a chromatogram of laboratory tap water using the method developed in this study. Panel A is the chromatogram from the conductivity detector. As shown in panel A, the most common anions in the water including fluoride, chloride, nitrite, bromide, nitrate, phosphate and sulfate were quickly coeluted in 4 minutes in the analysis. Divert flow to the MS started at 5.5 minutes, perchlorate was eluted at 7.4 min and detected by MS as shown in panel B. The presence of perchlorate (m/z 99 and 101) ions were monitored as shown in panel C. The area count ratio of m/z 99 to 101 in this analysis is 2.74, the value identified perchlorate elution. Perchlorate is quantified using the ions of m/z 101, 0.315 ppb of perchlorate was measured.

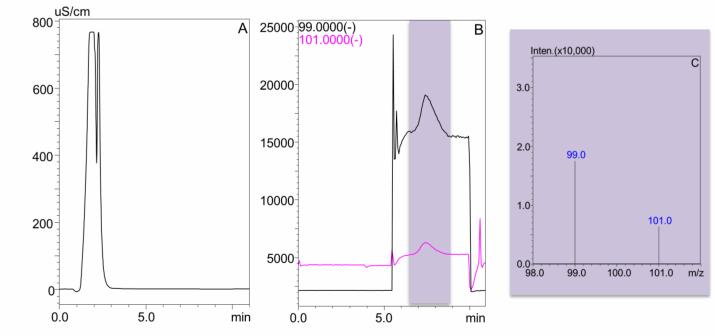


Fig. 3 Perchlorate analysis of tap water using IC-MS. Panel A: chromatogram from the conductivity detector, panel B: perchlorate chromatogram from MS, panel C: perchlorate (m/z 99 and 101) ions were found in mass spectra.

4. Conclusion

A Shimadzu suppressed anion chromatograph coupled with a single quadruple MS provides sensitive and reliable performance for the determination of perchlorate in drinking water. The method detection limit obtained in this study is 0.05 ppb, and the method precision, accuracy and sample recovery obtained in the method meets the quality criteria outlined by EPA 332.0.